

## Mechanics - Lecture 12 - 4/19/2022

[We'll start on 4/19 with the Lecture 11 notes.]

Big picture: (1) We discussed thus far the microcanonical ensemble, which uses the simple hypothesis that all pts on the set  $H = E$  are equally likely; so the assoc prob distr is "uniform measure on  $\{E < H < E + \Delta E\}$ ", i.e. the one s.t.

$$\langle f(q, p) \rangle = \frac{1}{Z_E} \int_{H=E} \frac{f(q, p)}{|dh|} dq dp$$

where  $Z_E$  is the normalization

$$Z_E = \int_{H=E} \frac{dq dp}{|dh|}$$

(2) But usually we observe a very limited system (e.g. a single particle) that's interacting with the rest of some larger system! We need a different prob distr for this. It is the canonical distribution, in which any energy  $E$  is permitted but likelihood is weighted by  $e^{-\beta E}$ ; here  $\beta$  is a constant, representing inverse temperature.

(3) Explaining the canonical ensemble is mostly about counting; we have no further use of the Hamiltonian eqns of motion. Therefore it can be done just as well for discrete systems as for continuous ones. This is a little easier, since "counting the no of states with energy  $E$ " is about actual counting rather than integration. I'll follow Buehler's book (2.3.2 + 3.3) and take this approach. (Tuckerman's book has the analogous ideas done directly in a continuous setting.)

Let's focus on the following very simple discrete system:

$$\begin{array}{ccccccc} \bullet & \bullet & \bullet & \bullet & \bullet & \bullet & \bullet \\ 1 & & & s_i = \pm 1 & & & N \end{array}$$

The "state" (a pt in phase space) is

$$s = (s_1, \dots, s_N), \text{ each } s_i = \pm 1$$

and the Hamiltonian is  $H(s) = \sum_{i=1}^N s_i$ .

(The famous 2D Ising model, which is a favorite example of a system with a phase transition, is not so different: there the "spins"  $s_i$  are on a 2D lattice and

$$H = -J \sum_{\text{neighbors}} s_\alpha s_\beta + B \sum_i s_i.)$$

In the microcanonical ensemble we treat each  $s = (\pm 1, \dots, \pm 1)$  as equally likely.

Viewing  $s_i$  as a coin flip, the statistics of  $H$  is very familiar:  $H = \sum_i s_i$  reports the # of heads - # tails for  $N$  flips. By Central Limit Theorem,  $s_1 + \dots + s_N$  is approx. Gaussian as  $N \rightarrow \infty$ , with mean 0 and variance  $N$ ; so

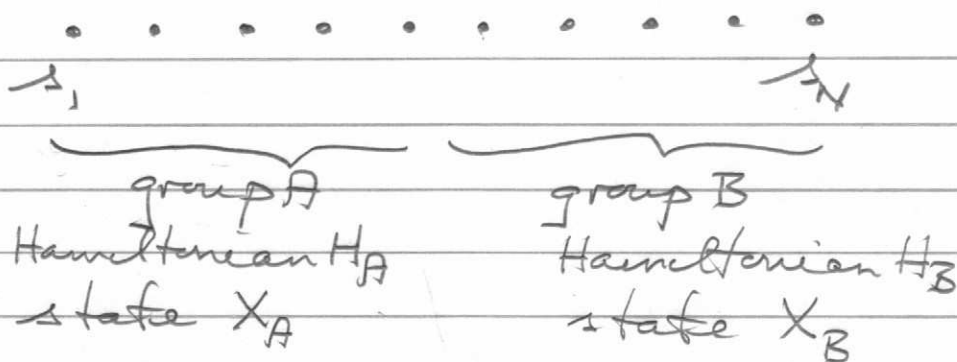
$$\begin{aligned} \Omega(E) &=_{\text{def}} \text{number of states with } H=E \\ &\approx C_N e^{-E^2/2N} \end{aligned}$$

when  $E$  is reasonably close to 0 &  $N$  is large. (While the exact formula is different from our calculation for  $N$  particles in Lecture 11, it shares the feature that  $\Omega(E)$  depends very strongly on  $E$ .)

It is convenient to introduce

$$S(E) = \ln \Omega(E)$$

which is called entropy. Obviously it measures the size of  $\Omega(E)$  - and therefore our uncertainty about the state if we know only that  $H = E$  - but so does  $\Omega(E)$  itself. To see why  $\ln \Omega(E)$  is useful, we do some warmup calculations involving what happens when we break the system into two parts



( $X_A$  = list of  $s_i$  in group A,  $H_A(X_A)$  = their sum, etc), Note that if  $X = (X_A, X_B)$  is state of entire system then  $H(X) = H(X_A) + H(X_B) = \sum_{j=1}^N s_j$ .

Question 1: Given that  $H(X) = E$ , what is the most likely value of  $E_A = H(X_A)$ ?

Ans: We want  $E_A$  that

$$\text{maximizes } \Omega_A(E_A) \cdot \Omega_B(E - E_A)$$

(since the A + B systems are independent),

Taking log,

$$\text{max}_{E_A} S_A(E_A) + S_B(E - E_A).$$

If  $N$  is large we can treat  $E_A$  as a  
cont's variable; then

$$\text{max is when } S'_A(E_A^*) = S'_B(E - E_A^*),$$

Defining the temperature  $T$  of a system  
by

$$S'(E) = \frac{1}{T} \quad \triangleright$$

we see that the answer to Question 1  
was: the most likely situation is  
when the two subsystems have  
the same temperature.

Note: we'll often be more interested  
in  $S'(E)$  than in  $T$ . So it is convenient

to work with

$$\beta = \frac{1}{T} = S'(E).$$

(Here entropy is nondimensional so  $T$  has units of energy. If we measure temp in degrees we need a conversion factor; that's Boltzmann's constant, usually written  $k_B$ , and then  $S'(E) = \beta = 1/k_B \cdot (\text{temp})$ .)

Question 2: Given that  $H(X) = E$ , find

a) the conditional probability of event  $X_A$ ?

b) the probability that  $H_A(X_A) = E_A$ ?

Ans to (a):

$$\text{Prob}(X_A | H=E) = \frac{\Omega_B(E - H_A(X_A))}{Z(E)}$$

where

$$Z(E) = \sum_{\substack{\text{all states} \\ X_A \text{ of system A}}} \Omega_B(E - H_A(X_A))$$

[In fact, if  $X_A$  is fixed then  $X_B$  must have  $H_B(X_B) = E - H_A(X_A)$  and all states with this energy are equally likely.]



Ans to (b):

$$\text{Prob}(H(X_A) = E_A | H = E) = \frac{\Omega_A(E_A) \Omega_B(E - E_A)}{Z(E)}$$

OK, now we're ready to think about a small system in equilibrium with a much larger one (a "heat bath"). We model this by keeping the size of A fixed but sending the size of B to  $\infty$ . (B is "the reservoir").

Start by rewriting answer to Question 2(e):

$$\text{Prob}\{X_A | H = E\} = \frac{1}{Z(E)} e^{S_B(E - E_A)}$$

where  $E_A = H_A(X_A)$ .

Use linear approx of entropy

$$S_B(E - E_A) \approx S_B(E) - S'_B(E) E_A$$

to see that

$$\text{Prob}\{X_A | H = E\} \approx c_E e^{-\beta E_A}$$

where  $\beta = S'_B(E)$

Evidently: holding  $E$  fixed, but focusing on system  $A$  and dropping the conditional probability notation,

$$\text{Prob of state } X_A = \frac{1}{Z(\beta)} e^{-\beta H_A(X_A)}$$

$$\text{where } Z(\beta) = \sum_{\substack{\text{states } X_A \\ \text{of system } A}} e^{-\beta H_A(X_A)}$$

This is the canonical distribution.

An important note: dropping  $A$  from our notation (since we're now only thinking about system  $A$ ),

The probability that  $H(X) = E$  under the canonical distribution is

$$\frac{1}{Z(\beta)} e^{-\beta E} \Omega(E) = \frac{1}{Z(\beta)} e^{[S(E) - \beta E]}$$

so the most likely value of  $E$  is the one that minimizes  $E - \frac{1}{\beta} S(E) = E - TS(E)$ . This is the "free energy". (Evidently:



a value  $E$  becomes more likely either because it is lower ( $\Rightarrow e^{-\beta H}$  is larger) or because it is achievable many different ways (larger entropy).

Bottom line: at finite temperature the most likely value of  $H$  is determined by minimizing the free energy